

Model systems for photoelectrochemical water splitting: A surface science approach

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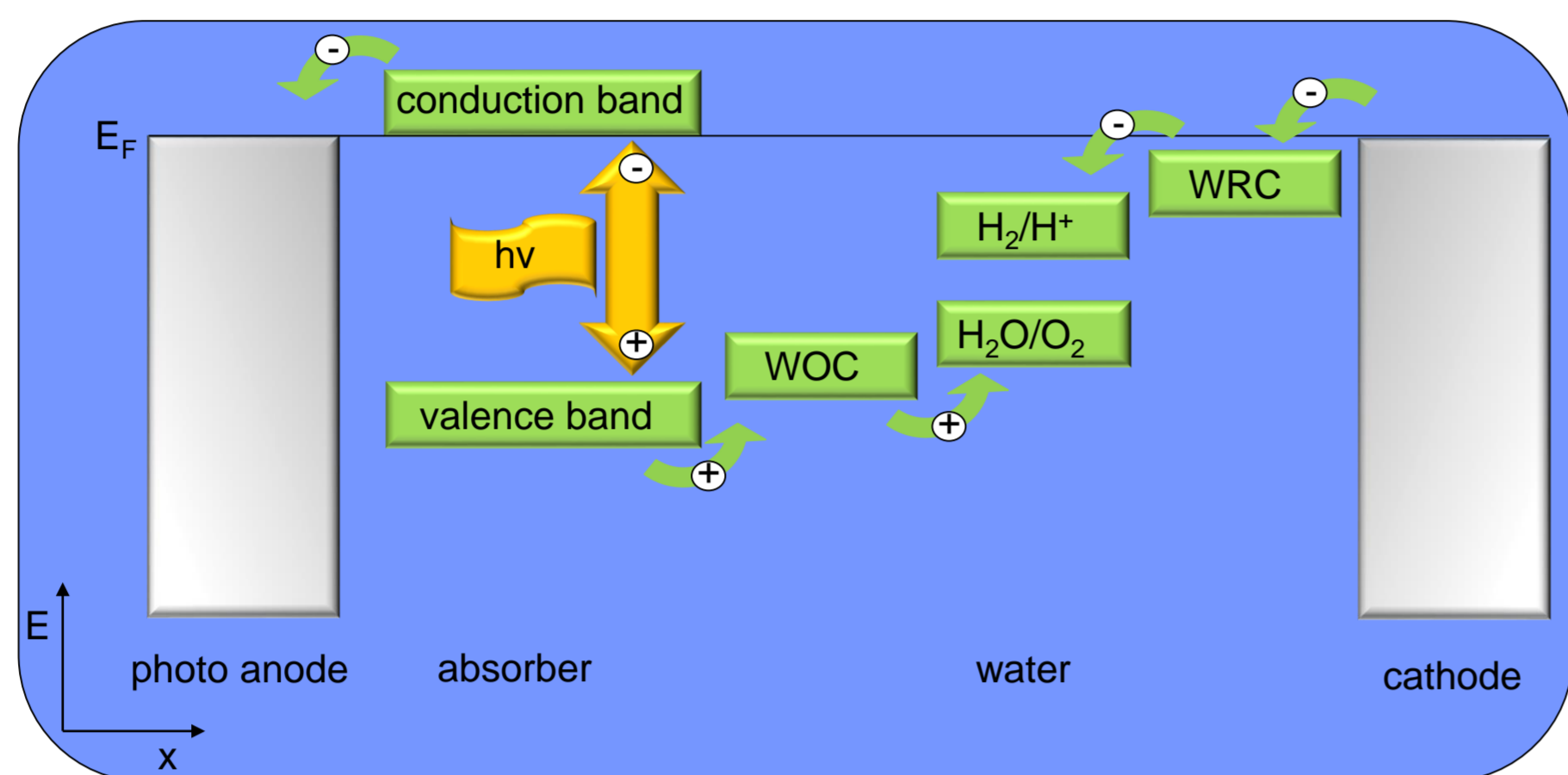
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1. Motivation

- water splitting as approach for energy storage of solar light: $2\text{H}_2\text{O} + h\nu \rightarrow 2\text{H}_2 + \text{O}_2$
- H_2 as fuel of the future
- model systems for water splitting photoelectrodes
- mechanistic insight about novel catalyst molecules, photosensitizers and interfaces

Working principle of a photoelectrochemical cell:

- solar photon is absorbed and creates an electron-hole pair
- possible absorbers: semiconductors or photosensitizers
- charge transfer to water reduction catalyst (WRC) and water oxidation catalyst (WOC)
- dissociation reaction of water:
 $4\text{H}^+ + 4\text{e}^- \rightleftharpoons 2\text{H}_2$
 $2\text{H}_2\text{O} + 4\text{h}^+ \rightleftharpoons 4\text{H}^+ + \text{O}_2$



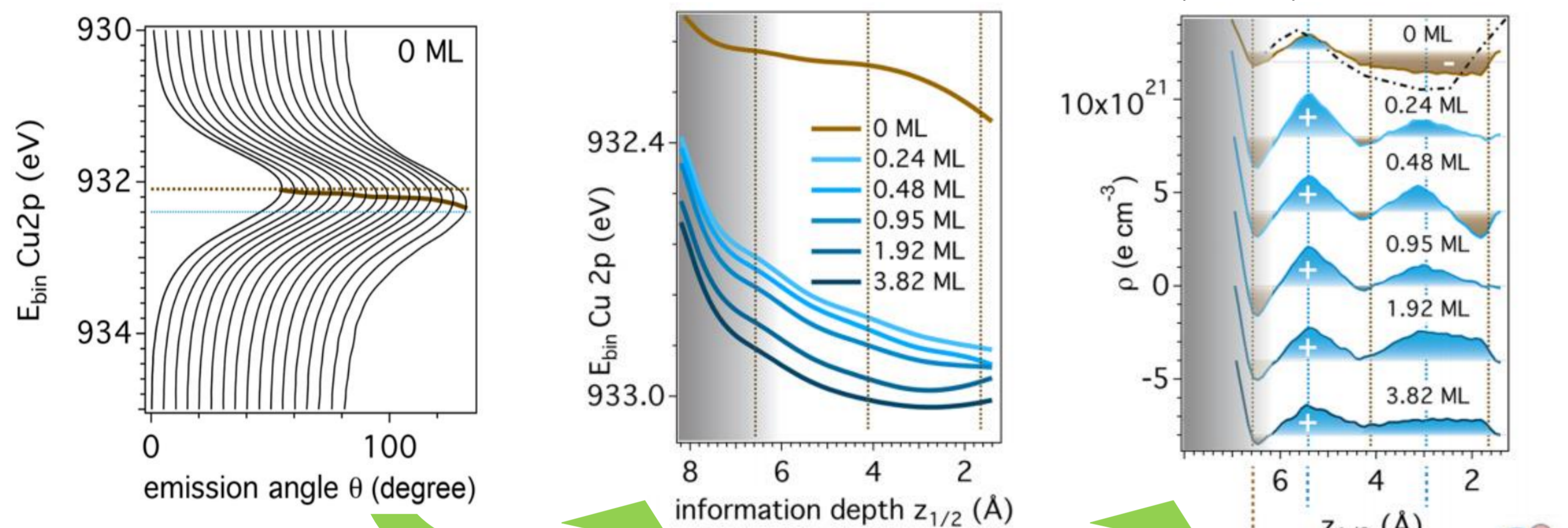
2. Atomically Resolved Band Bending Effects

- heterojunctions of semiconductors and molecules of importance
- angle-resolved x-ray photoelectron spectroscopy (XPS) provides spatial resolution of band bending
- atomically resolved band bending allows extraction of charge density

XPS of the Cu 2p_{3/2} core level as a function of emission angle for Cu₂O(111)

peak maximum of Cu 2p_{3/2} core level as a function of information depth

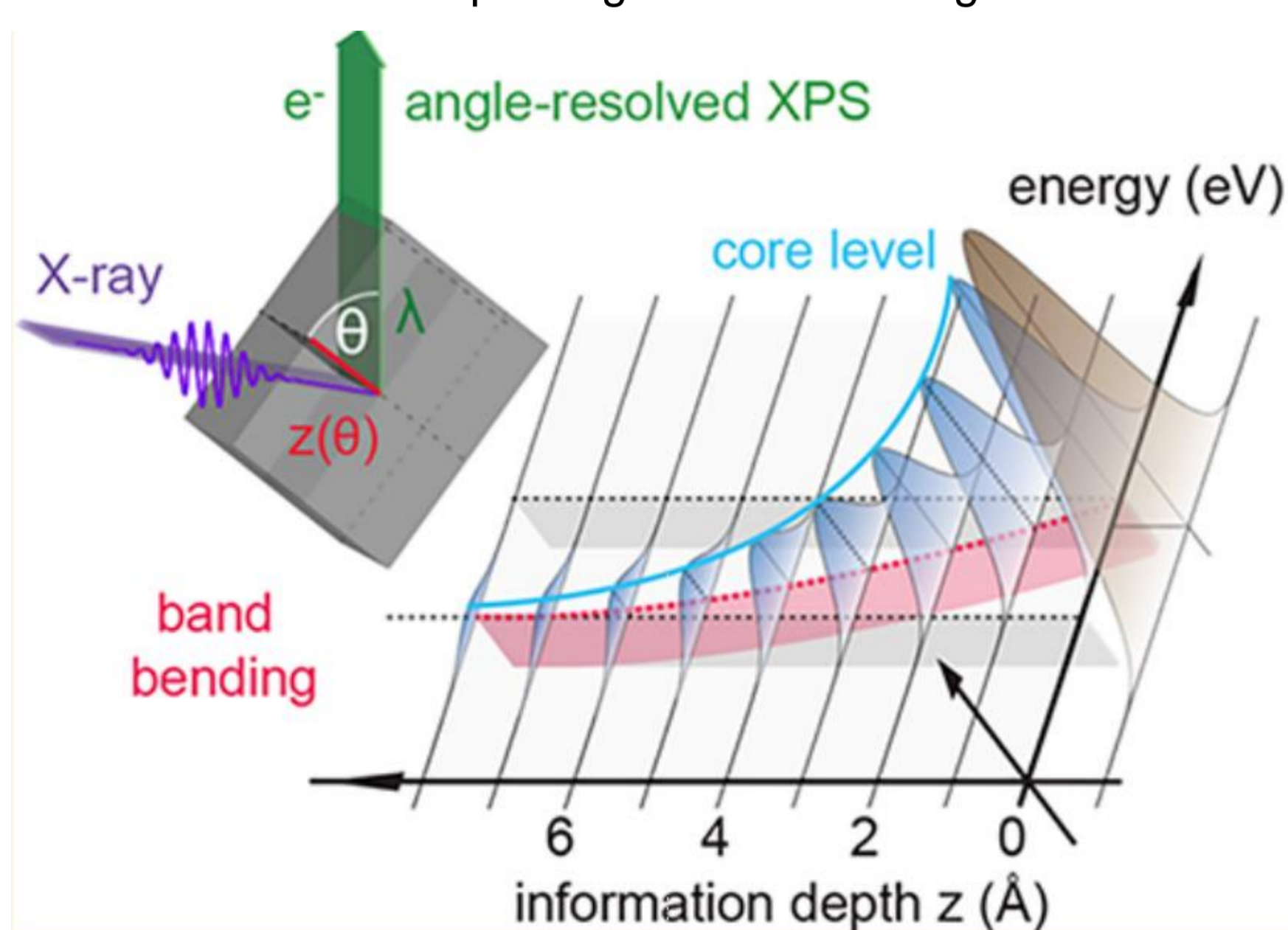
extracted charge density profiles $\rho(z)$ perpendicular to the (111) surface (top) and structural model (bottom)



$$z_{1/2}(\theta) = \ln(2)\lambda_{in}(E_{kin})\cos(\theta)$$

$$\rho(z) = -\epsilon_0\epsilon_r \frac{d^2 E_{bin}}{dz^2}$$

working principle of angle resolved XPS: spatial resolution allows probing of band bending



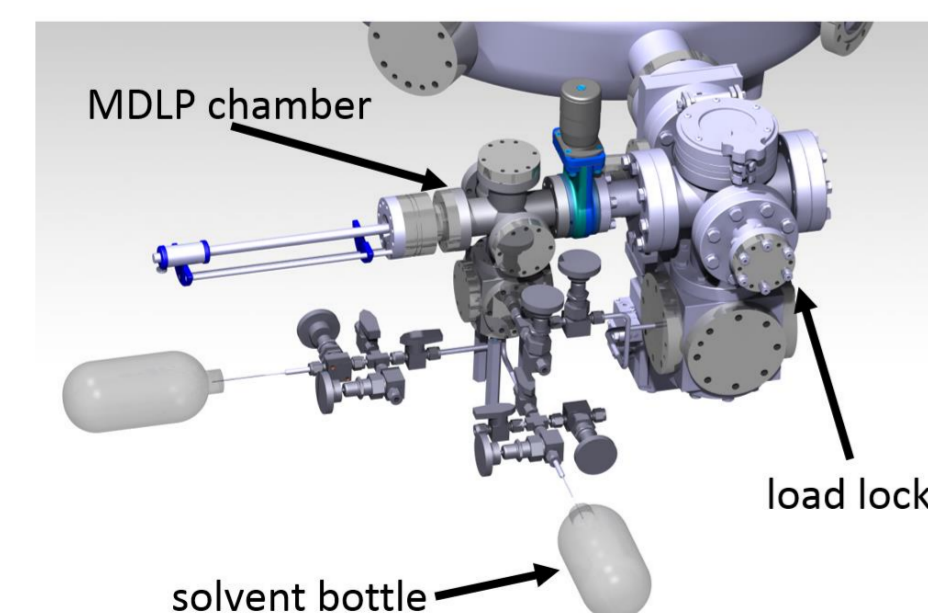
used water reduction catalyst: Co-Pyrrophyrin (CoPyr)



Leuenberger et al., *Nano Lett.*, 17 (2017) 6620

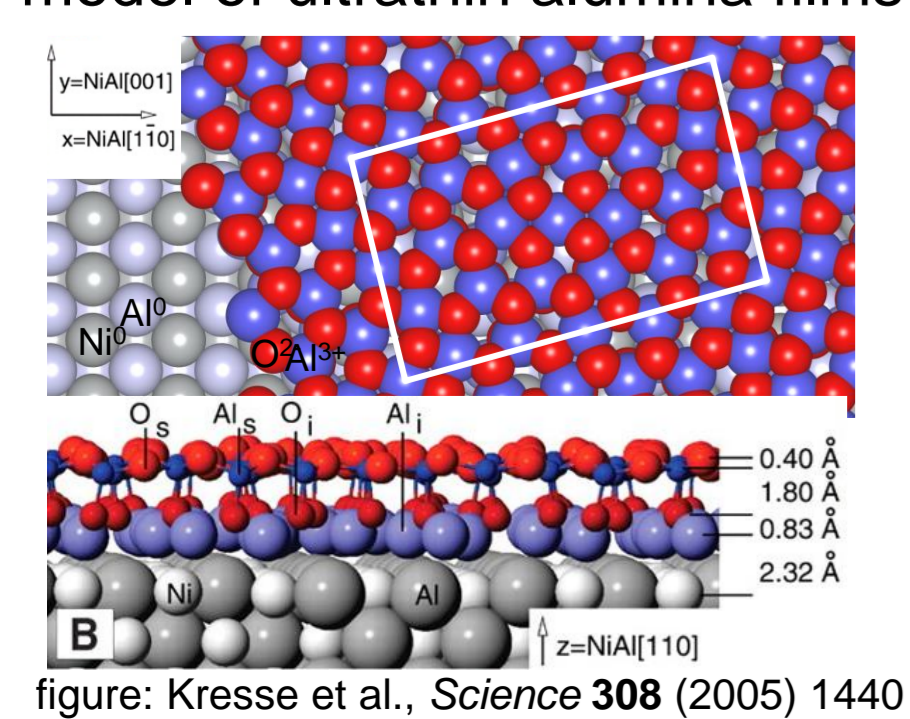
3. Wet-chemical Functionalization of Ultrathin Films

CAD model of new setup

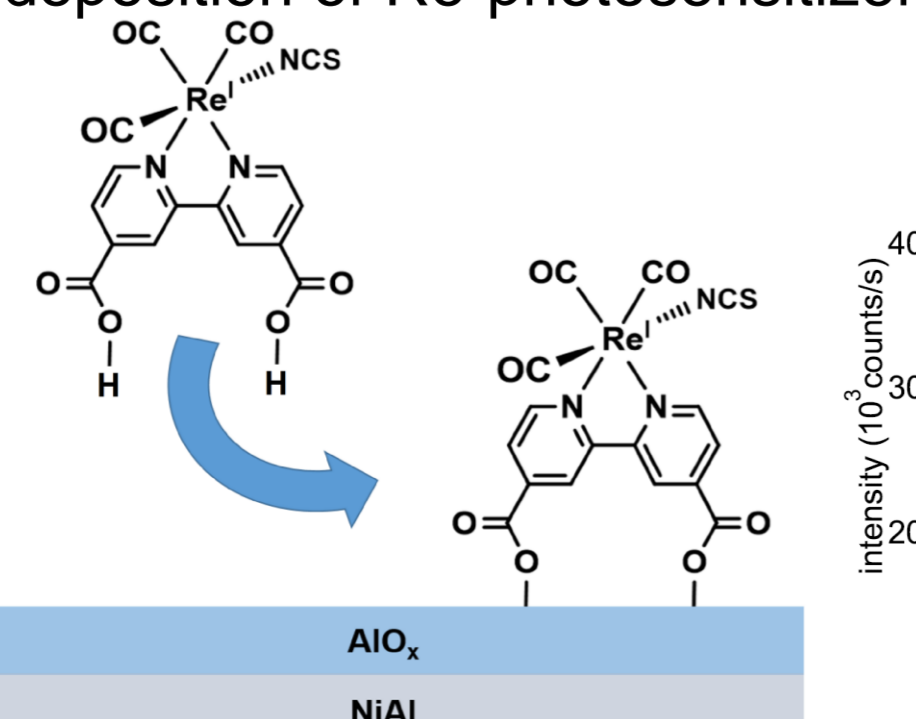


- ultrathin alumina films prepared under UHV-conditions
- surface oxide not stable in air [3]
- new chamber developed for wet-chemical deposition of molecules
- high vacuum cleanliness maintained
- Re(NCS)(CO)₃(dicabipy) as photosensitizer:
 - molecule density: $n_{Re} = 1.0 / \text{nm}^2$
 - molecules passivate surface

model of ultrathin alumina films



deposition of Re-photosensitizer



XPS after molecule deposition (red), and after 5 minutes of air exposure (black). The inset compares mono- (blue) and multilayer (red).

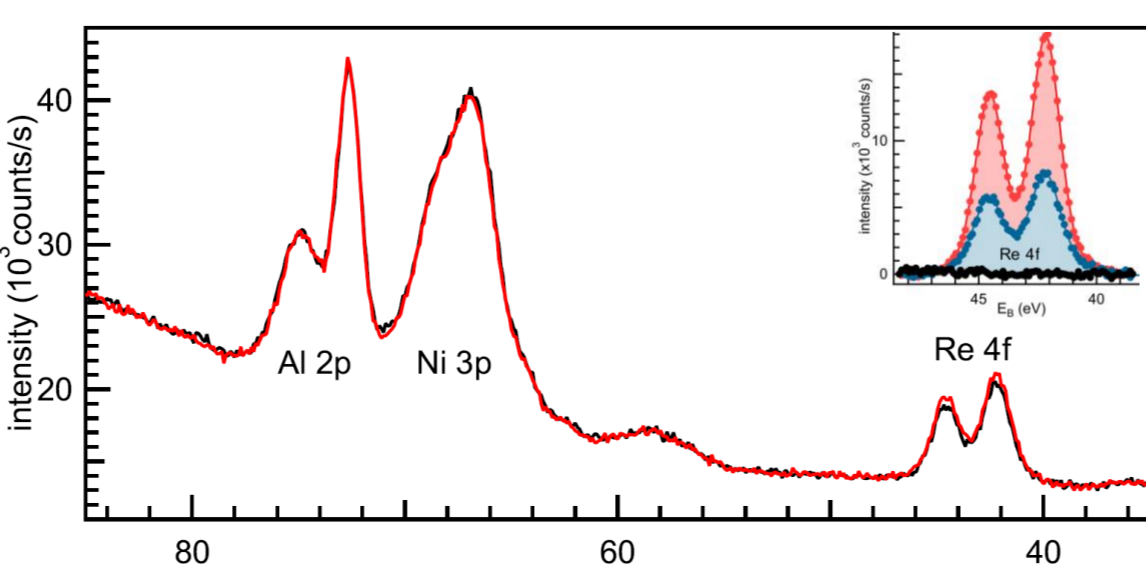
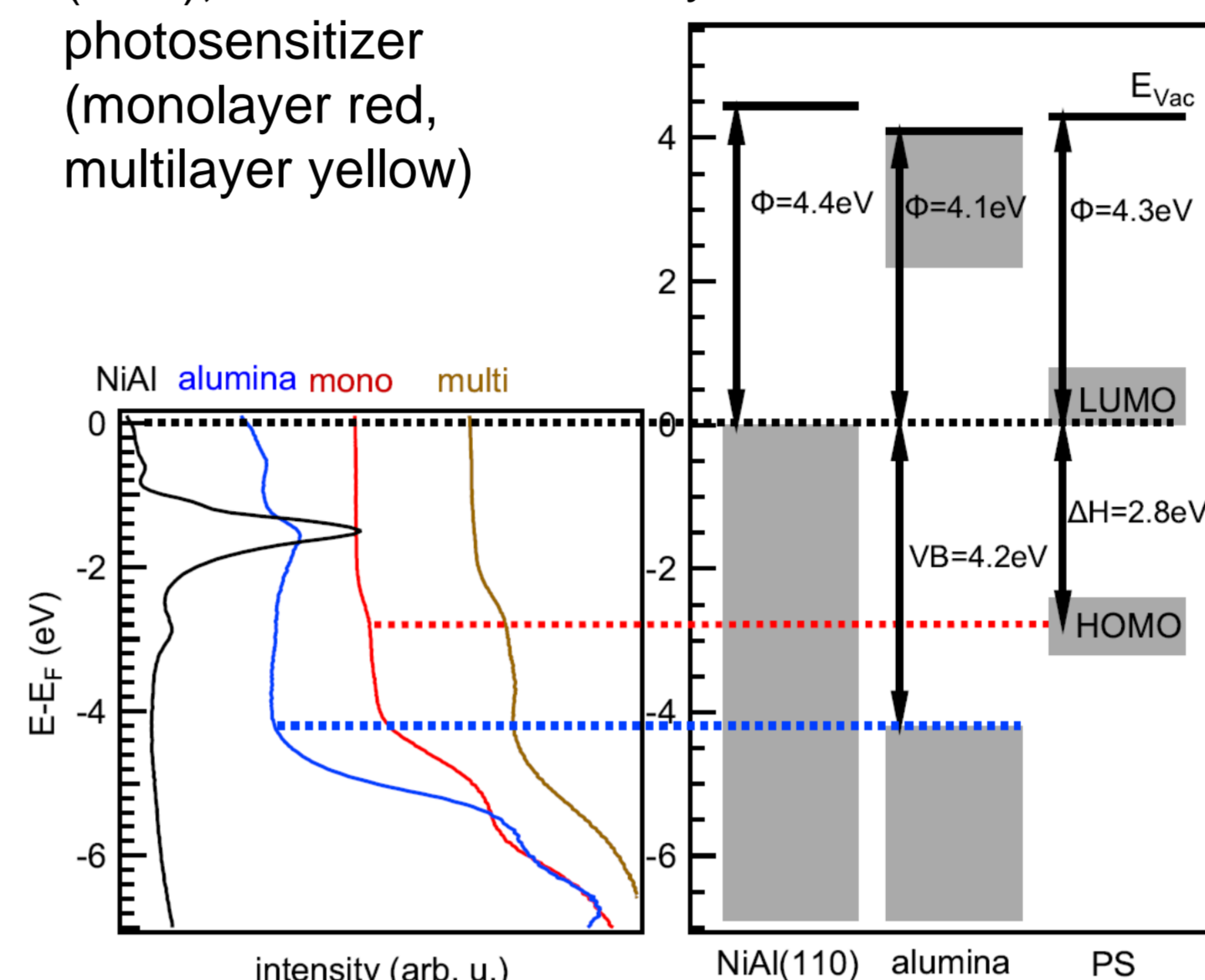


figure: Kresse et al., *Science* 308 (2005) 1440

4. Energy Alignment and Modulation of Charge Carrier Dynamics

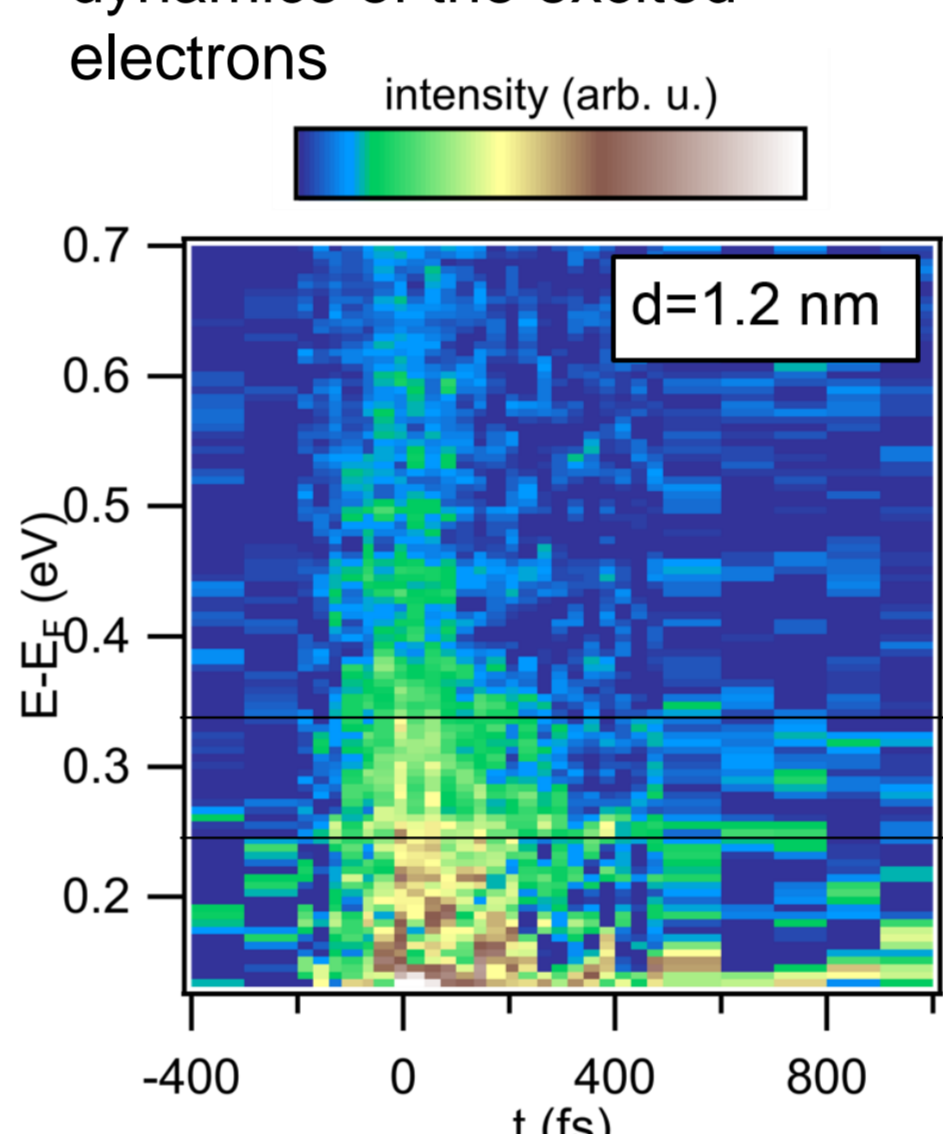
UPS of NiAl substrate (black), alumina film (blue), and Re-photosensitizer (monolayer red, multilayer yellow)

energy alignment in photoelectrode model system

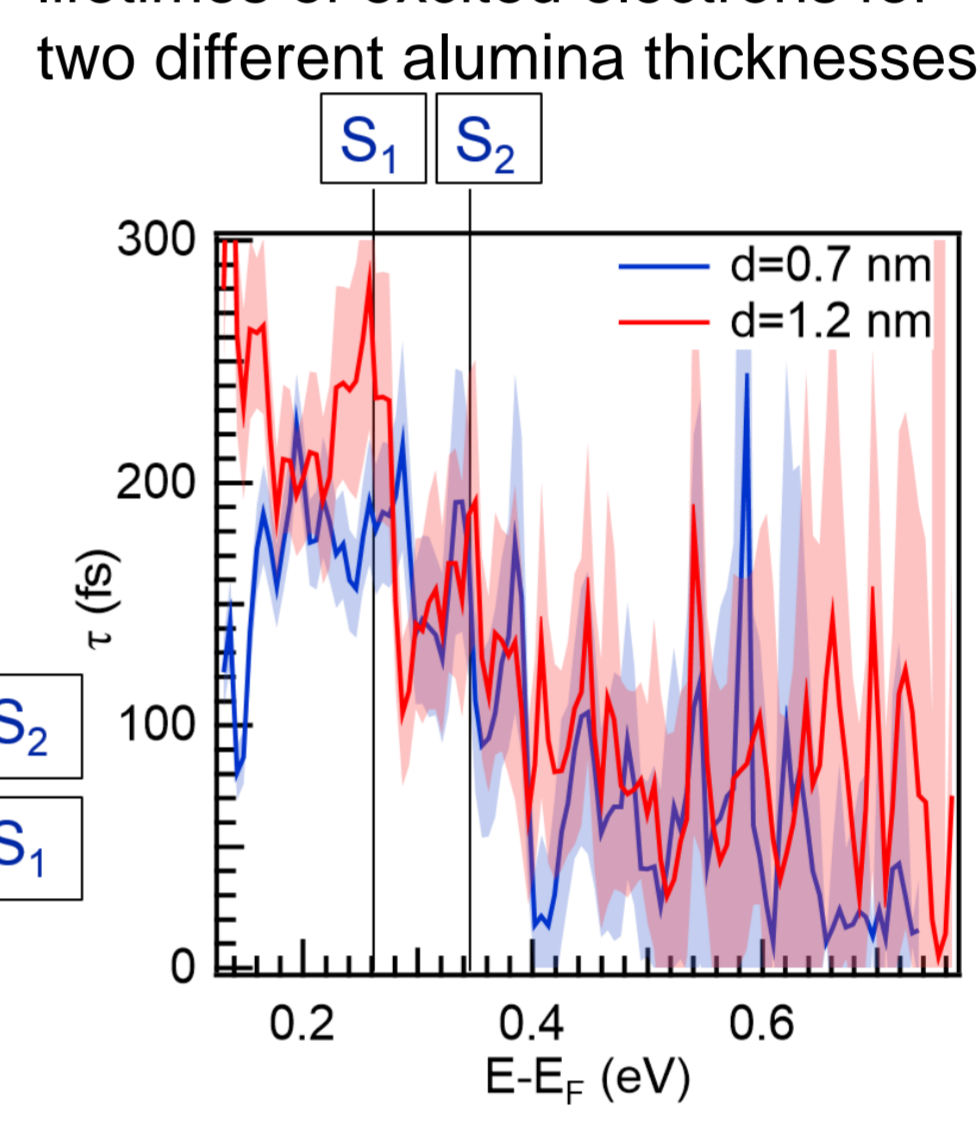


- ultraviolet photoelectron spectroscopy (UPS) from substrates and molecular overlayers
- information about valence band structure, occupied molecular orbitals, ionization potentials and energy alignment
- time-resolved two photon photoemission for electron dynamics (pump-probe experiment)
- lifetime of excited electrons in photosensitizer changes for different alumina thicknesses
- electron injection from molecule to substrate for thinner films
- tunneling relevant process in such devices

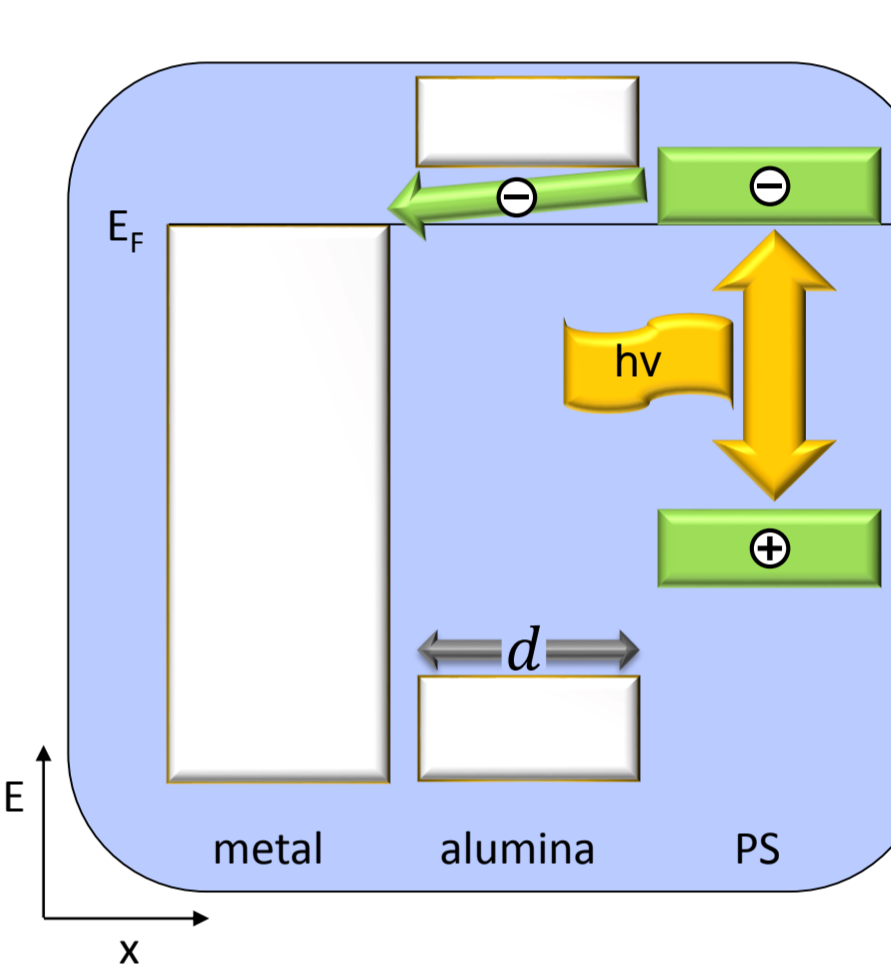
dynamics of the excited electrons



lifetimes of excited electrons for two different alumina thicknesses



excited electrons in the photosensitizer are injected into the metal substrate



5. Instrument Development: Solid-Liquid Ambient Pressure XPS

- new endstation attached to existing Scienta HiPP2 analyzer at Paul Scherrer Institute
- in-situ UHV sample preparation capabilities
- solid-liquid interface created by the dip-and-pull method [4]
- commissioning scheduled for June 2018

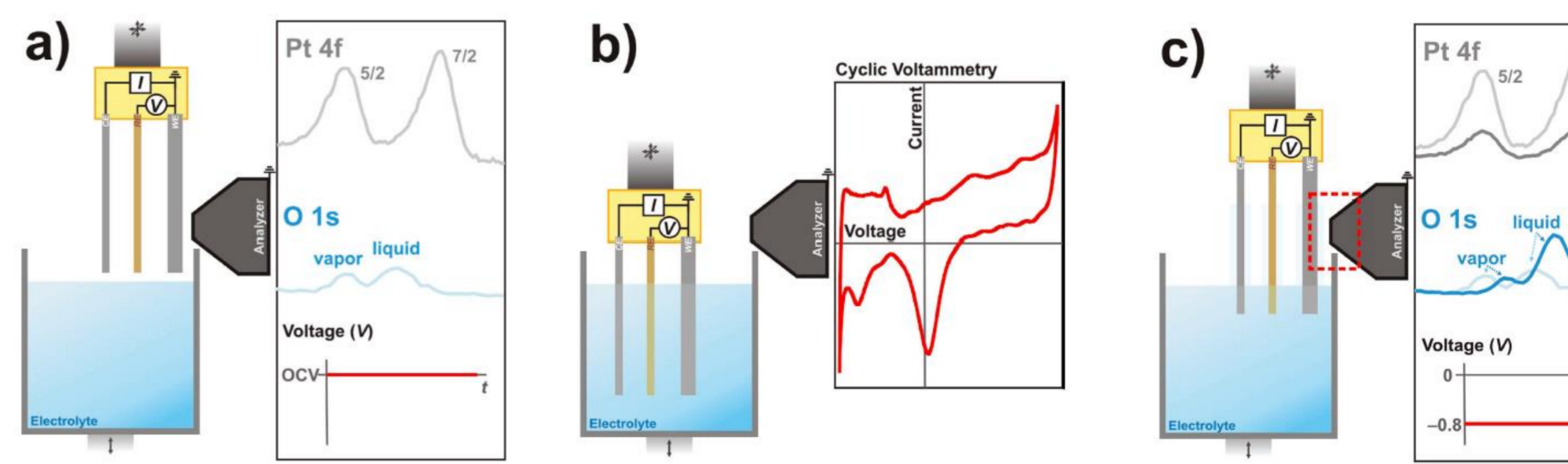


In situ electrochemical cell

Water vapor pressure ~ 25 mbar



$h\nu = 4 \text{ keV}$



Prior to immersion: O 1s shows signals from water vapor and from a condensed water film.

During immersion: Measure cyclic voltammograms, Sample can be cleaned.

After immersion: Liquid water film pulled out of the liquid, Potential control of liquid water and vapor.

Axnanda et al., *Sci. Rep.*, 5 (2015) 9788

Outlook

- ongoing effort towards studying molecular catalysts and photosensitizers under catalytically relevant conditions
- understanding electron dynamics in model photoelectrodes
- implementing time-resolved corelevel spectroscopy to probe metal-to-ligand charge transfers

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