

PHY127 FS2024

Prof. Ben Kilminster
Lecture 9
May 3rd, 2024

Penetration of X-rays

combined effect of Thomson scattering, photoelectric effect & compton scattering generate attenuation of the X-ray beam.

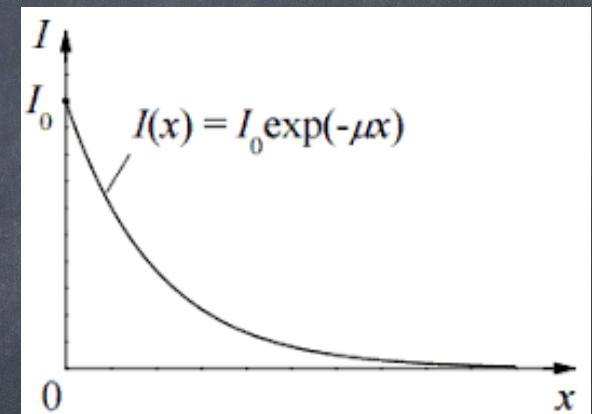
$$I(x) = I_0 e^{-\mu x}$$

I_0 : initial beam intensity,

$I(x)$: intensity at a depth, x

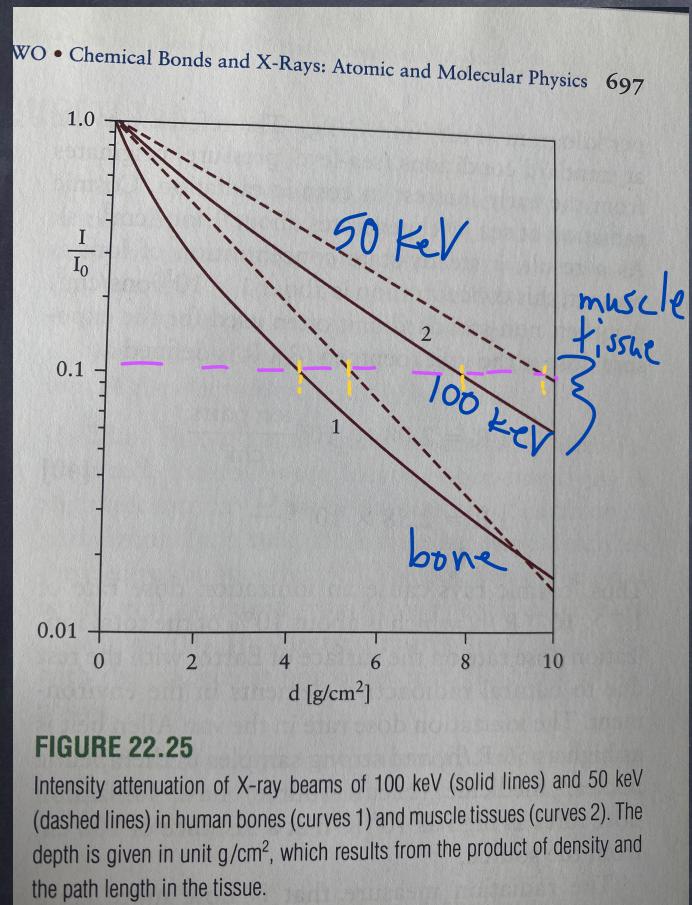
μ : attenuation coefficient
with units $[m^{-1}]$

The mass attenuation coefficient, μ/ρ ,
is the attenuation per unit density
of the material being penetrated.
Units : $\left[\frac{cm^2}{g}\right]$



photon energy	$\frac{\mu}{\rho_{water}}$	$\frac{\mu}{\rho_{dry air}}$	$\frac{\mu}{\rho_{bone}}$	$\frac{\mu}{\rho_{muscle}}$	$\frac{\mu}{\rho_{breast tissue}}$
100 keV	0.17	0.15	0.18	0.16	0.15
10 keV	5.3	5.1	28	5.3	4.3
5 keV	43	40	190	47	34

Observations: The higher the γ -ray energy, the farther the γ -rays penetrate.
 (If μ is large, attenuation is more & the distance traveled is less)



The τ -axis is given as the product of density ρ and the path length, x . $d = \rho \cdot x$
 (Because the combination is more meaningful)
 For instance, γ -ray intensity is reduced to 10% ($I/I_0 = 0.1$)...
 for muscle tissue, at $d = 8-10 \frac{\text{g}}{\text{cm}^2}$

$$x = \frac{d}{\rho_{\text{mus}}} \approx \frac{9 \frac{\text{g}}{\text{cm}^2}}{1 \frac{\text{g}}{\text{cm}^3}} = 9 \text{ cm}$$

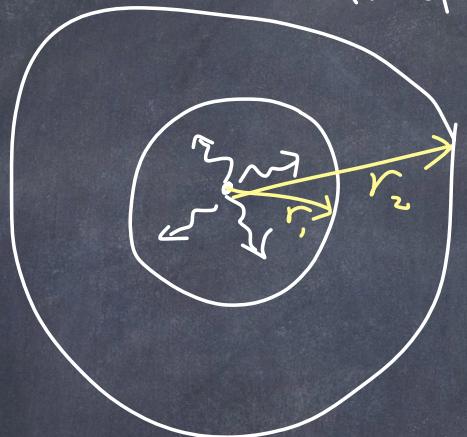
 for bone tissue, $d = 4.5-5.5 \frac{\text{g}}{\text{cm}^2}$

$$x = \frac{d}{\rho_{\text{bone}}} \approx \frac{5 \frac{\text{g}}{\text{cm}^2}}{1.2 \frac{\text{g}}{\text{cm}^3}} \approx 4 \text{ cm}$$

From [1]: "Physics of the Life Sciences" by Martin Zinke-Allmang

Reminder:

$$\text{Intensity} = \frac{\text{Power}}{\text{area}}$$



Intensity will decrease like $\frac{1}{r^2}$
Surface Area of a sphere is $A=4\pi r^2$

If you have $r_2 = 2r_1$,
the intensity is 4 times less
at r_2 than r_1

Where does the t-ray intensity go?

t-rays are either scattered or absorbed by bone or tissue.

Absorbed radiation has an adverse biological effect.

Measured radiation is reported in 2 ways:

1) amount of ionization occurring in the material due to the radiation → exposure dose

2) energy deposited by radiation in the material
→ absorbed dose

Dose = total amount of ionization or energy deposited in a given amount of material.

Dose rate = dose per unit time.

There are different measures:

exposure dose: total charge generated by ionization per kg of air
units: $\left[\frac{\text{C}}{\text{kg}} \right]$ std. atmosphere

Cosmic radiation at sea level generates $1-10 \text{ ions/cm}^3/\text{s}$. This equilibrium value is maintained despite recombination of ions.

Another unit is roentgen (R)

$$1 R = 2.08 \times 10^9 \frac{\text{ion pairs}}{\text{cm}^3}$$
$$= 2.58 \times 10^{-4} \frac{\text{C}}{\text{kg}}$$

So cosmic rays cause ionization dose rate of $1.7 \times 10^{-6} \text{ R/h}$. This is about 10% of dose rate at earth's surface. The other 90% is natural radioactivity (Radon, ... ^{41}K (bananas))

More commonly used is the energy dose, the energy deposited per kg of air in units of $\left[\frac{\text{J}}{\text{kg}} \right]$
Units are called "gray"

$$1 \text{ gray} = 1 \text{ Gy} = 1 \frac{\text{J}}{\text{kg}}$$

Sometimes an older unit is the "rad"

$$1 \text{ Gy} = 100 \text{ rad}$$

$$1 \text{ R} \cong 1 \text{ rad} = 0.01 \text{ Gy}$$

Biological effect of radiation
defined as equivalent dose,

D_{equivalent}, with units of Sievert (Sv)

Defined so that the same value of D_{equivalent} has the same impact on living tissue, for any type of radiation.

$$D_{\text{equiv}} = w_R \cdot w_T \cdot D_{\text{absorbed}}$$

radiation Factor,
expresses the physiological
damage relative to γ -ray
radiation:

$$w_R = 1 \quad \text{For } \gamma\text{-rays, electrons,}\br/> \text{positrons}$$

$$w_R = 5-10 \quad \text{for neutrons}$$

$$w_R = 10 \quad \text{alpha particles (He nucleus)}$$

w_T: tissue weighting factor
for whole body, this sums to 1.
with respect to a whole body
exposure.

D_{absorbed}: radiation exposure
in Gray.

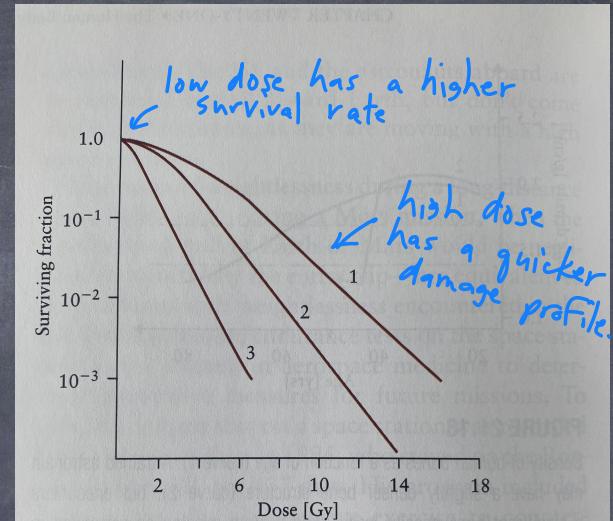


FIGURE 21.15

The surviving fraction of three types of human cells as a function of energy dose in unit Gy. The energy dose is the energy deposited by the radiation per kilogram of tissue. Note the lower steepness at doses below 1 Gy, which is due to self-repair mechanisms in living cells. Various cells respond with different sensitivity to radiation: (1) thyroid cells, (2) mammary cells, and (3) bone marrow.

[1]

Tissue weighting factors from ICRP

Effect of dose

Equivalent dose (Sv)

1 - 5

4 - 5

10 - 50

50 - 100

L

↑
For acute dose (all at once)

pathological diagnosis

serious temporary alterations of blood count

50% death rate in 30 days.

vomiting + nausea (die sooner)
brain + nerve damage
(death in ~1 week)

	Female	Male
Testes	0	0.08
Ovaries	0.08	0
Bone surface	0.01	0.01
Bladder	0.04	0.04
Bone marrow, red	0.12	0.12
Brain	0.01	0.01
Breast	0.12	0.12
Colon	0.12	0.12
Liver	0.04	0.04
Lungs	0.12	0.12
Oesophagus	0.04	0.04
Salivary glands	0.01	0.01
Skin	0.01	0.01
Stomach	0.12	0.12
Thyroid	0.04	0.04
Remainder ^a	0.12	0.12

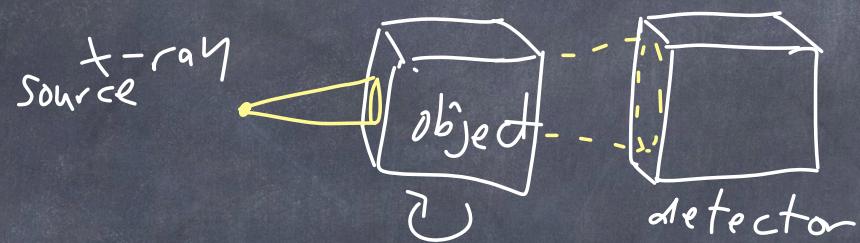
^aComponent organs for remainder in ICRP 103: adrenals, extrathoracic airways, gallbladder, heart, kidneys, lymphatic nodes, muscle, oral mucosa, pancreas, prostate, small intestine, spleen, thymus and uterus/cervix.

↑
sum should add up to 1.

X-ray tricks for medical use

- 1) gastrointestinal tract can be imaged by X-rays if filled with dense Barium ($\rho = 3.5 \frac{g}{cm^3}$) solution for increased contrast
- 2) similarly, iodine ($\rho = 4.93 \frac{g}{cm^3}$) + water, make a soluble organic compound, used for cardiovascular system, urinary tract, + the brain
- 3) mammography (lower energy X-rays, softer)
- 4) Improved images with Computed tomography (CT) to obtain 3-D images from a collection of 2-D images (X-ray images)

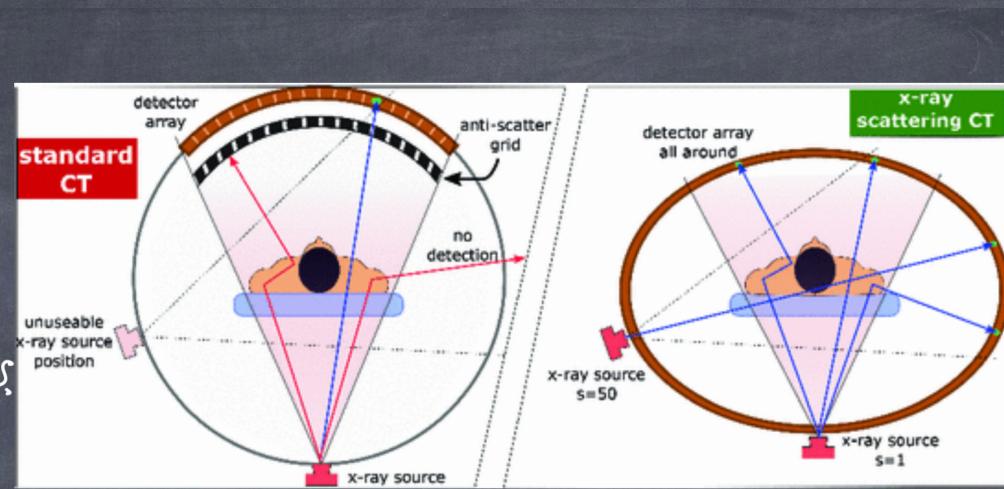
Originally, CT scan would have a single x-ray source, and a detector opposite.



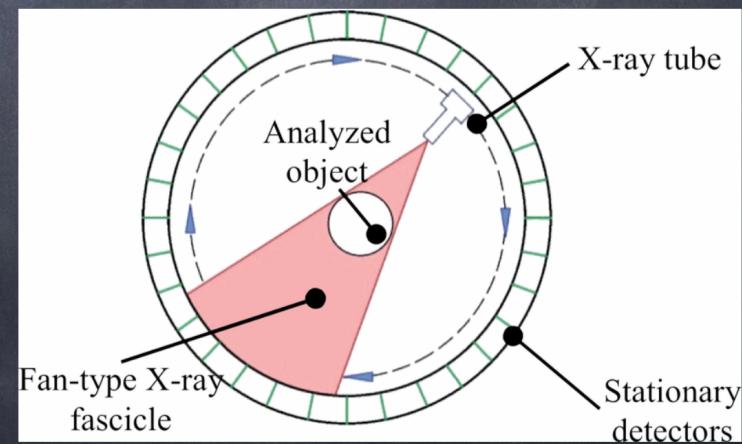
Rotate by 1° , take another x-ray.
~ minutes to do x-ray.

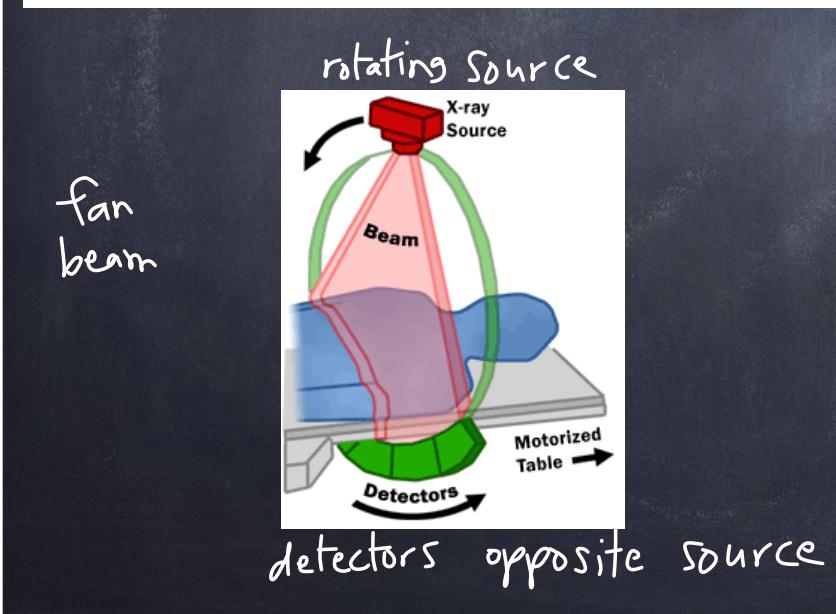
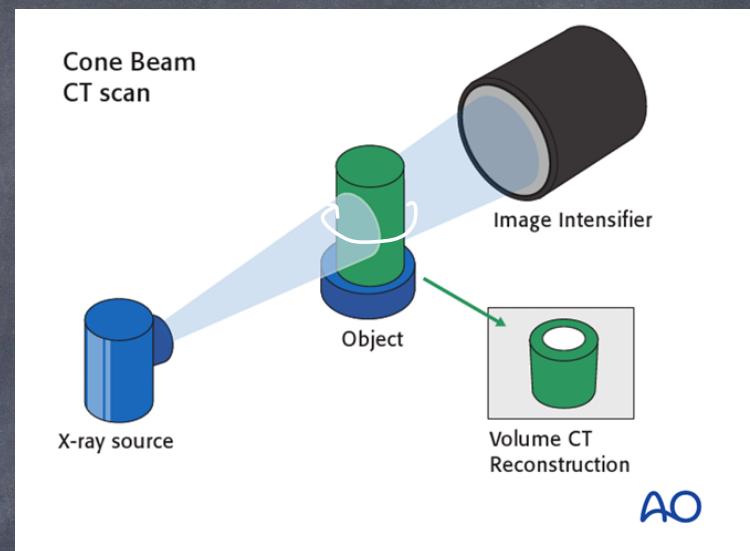
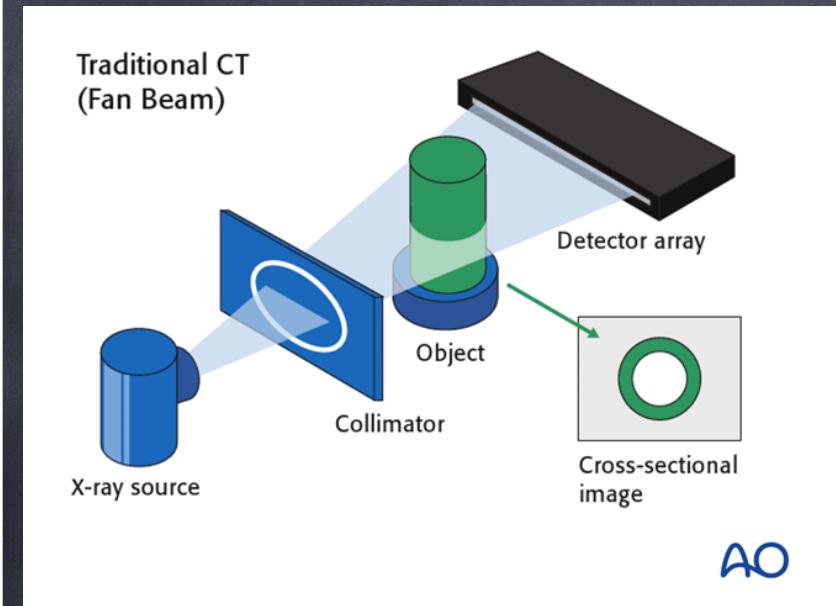


Today, wide fan-like beam, hundreds or thousands of detectors,
⇒ decreases time down to a few seconds



Newest: stationary detectors, and the beam sweeps around the patient.
Typically 50 ms per angle.





Gray scale can be converted into color to represent brightness level, set according to absorption coefficient of tissue, μ , compared to water, μ_w :

$$\text{Hounsfield unit} \quad HU = 1000 \frac{\mu - \mu_{\text{water}}}{\mu_{\text{water}} - \mu_{\text{air}}} \quad \text{OR}$$

$$\text{CT number} = 1000 \frac{\mu - \mu_w}{\mu_w}$$

Note:
Since μ_{air} is 800 times smaller than μ_{water} , $\text{CT number} \approx HU$

material

water

air

bone

muscle

fat

CT number for 60 KeV x-rays

0

-1000

808

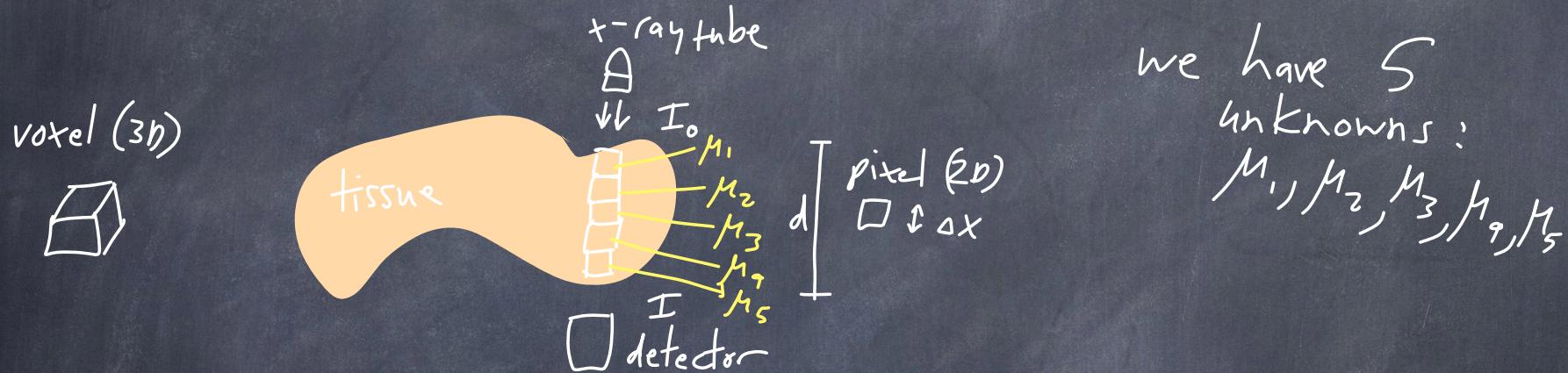
-48

-142



How to do and reconstruct a CT scan

X-ray beam goes through patient, different types of tissue are encountered, with different μ .



we have 5

unknowns:

$$\mu_1, \mu_2, \mu_3, \mu_4, \mu_5$$

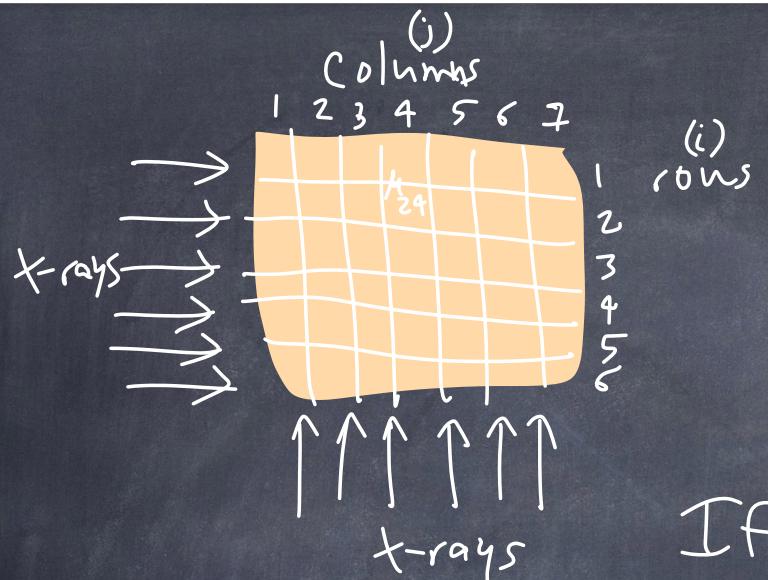
$$I = I_0 e^{-\sum_{i=1}^5 \mu_i \Delta x}$$

↑ we know this

$$\log \frac{I_0}{I} = \sum_{i=1}^5 \mu_i \Delta x$$

IF we keep making Δx smaller ($\Delta x \rightarrow 0$), then

$$I = I_0 e^{-\int_0^d \mu(x) dx}$$



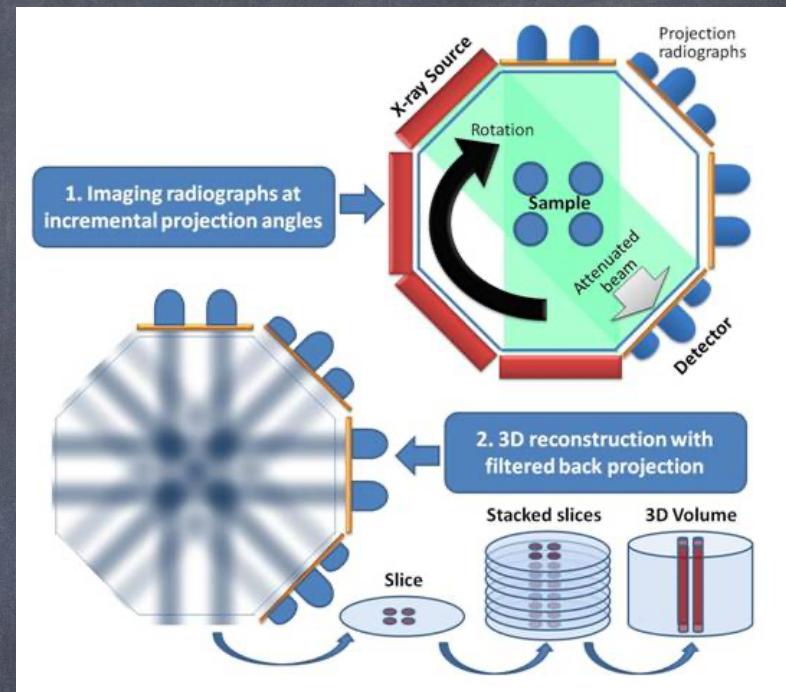
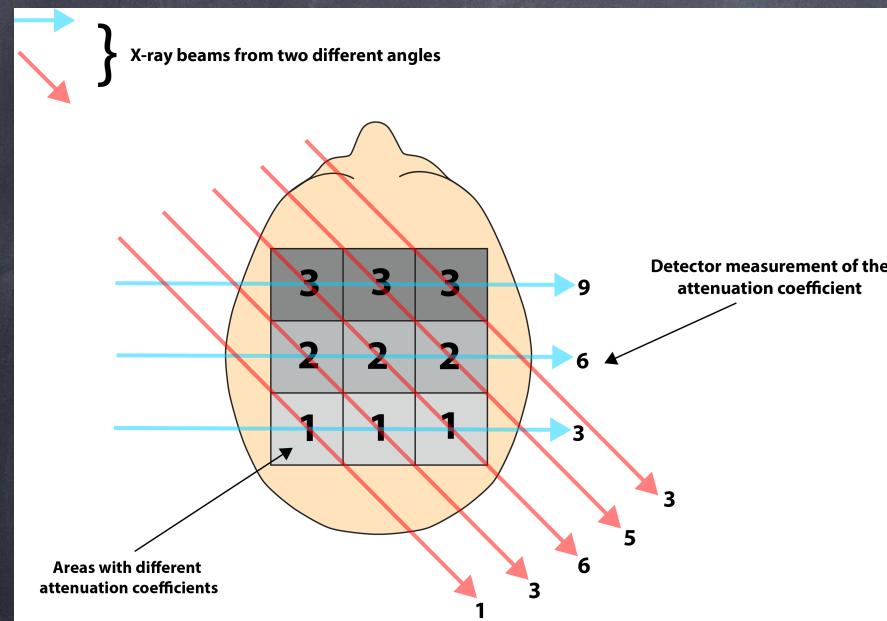
we can measure I along different rows & columns

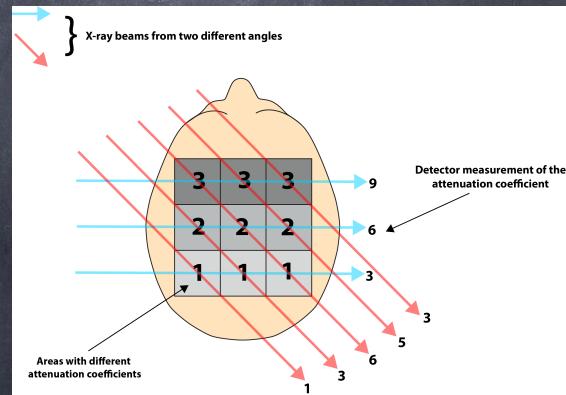
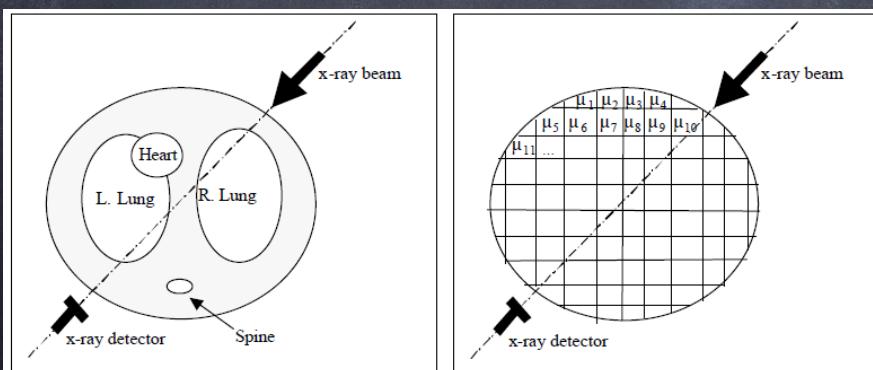
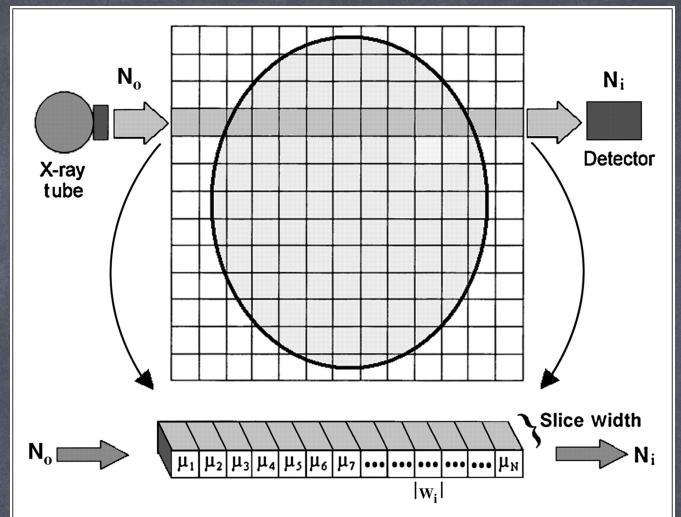
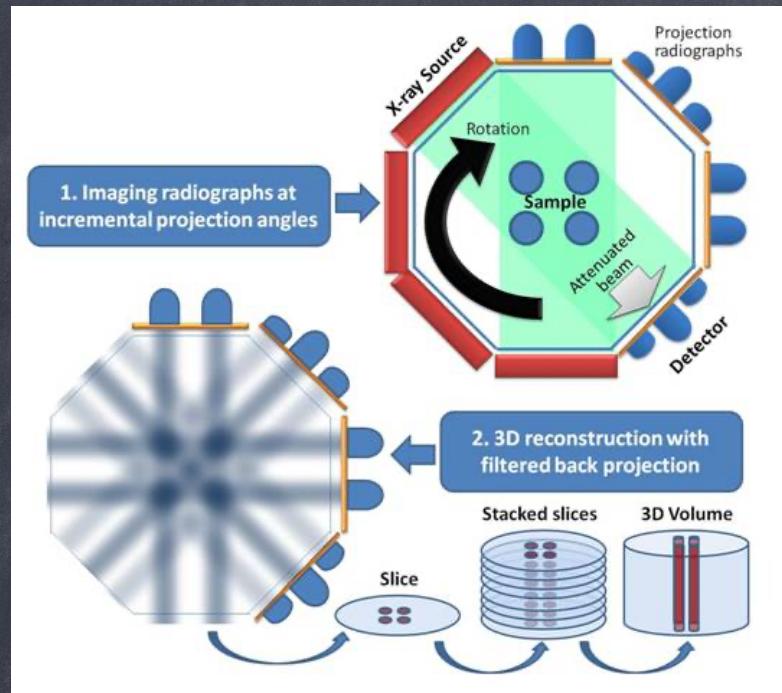
For the i^{th} row :

$$M_i(\Delta x) = \sum_{j=1}^7 M_{ij} \Delta x$$

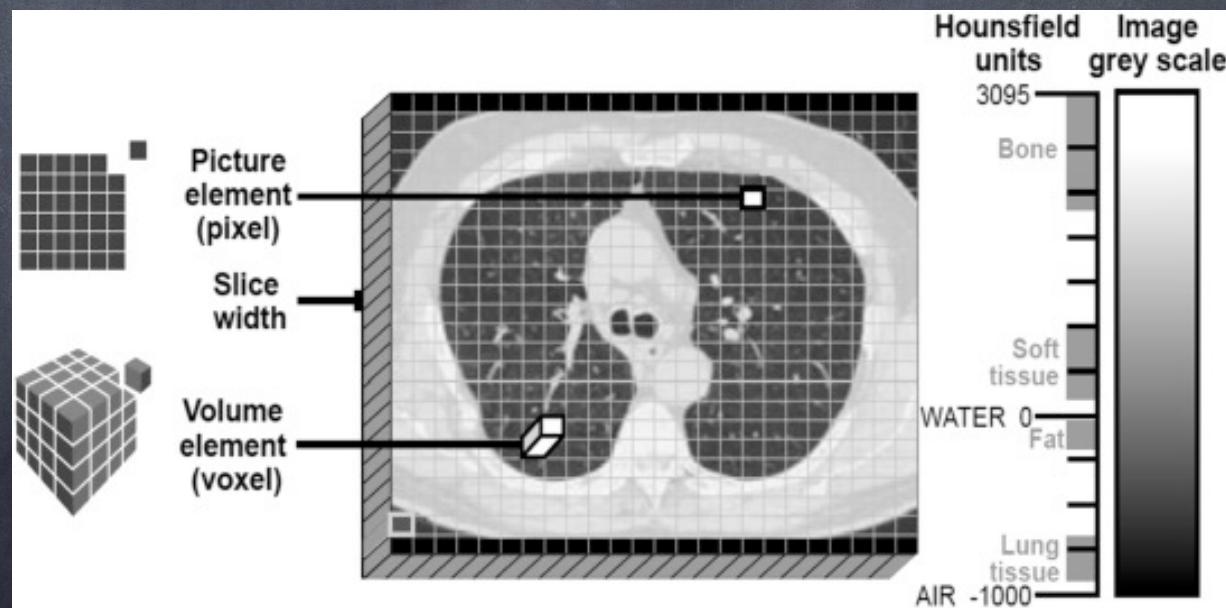
If we have N^2 pixels ($N \times N$ grid)
 here 7×7
 then you have N^2 unknowns, need N^2 equations to solve for the unknowns.

$2D \rightarrow 3D$

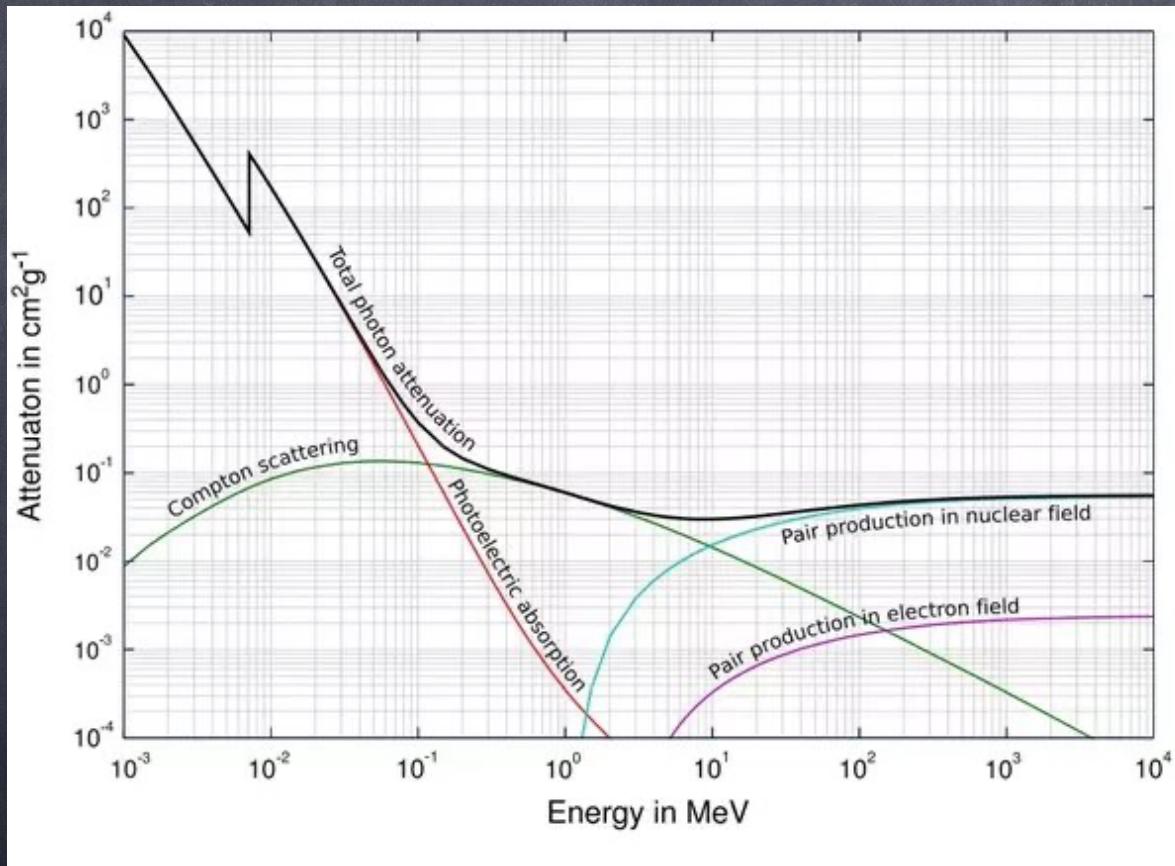




Typically, N : $256 \sim 1024$
 N^2 (#pixels) \sim 1 million unknowns.
Typically solved by computers



Reminder of attenuation of γ -rays + gamma rays

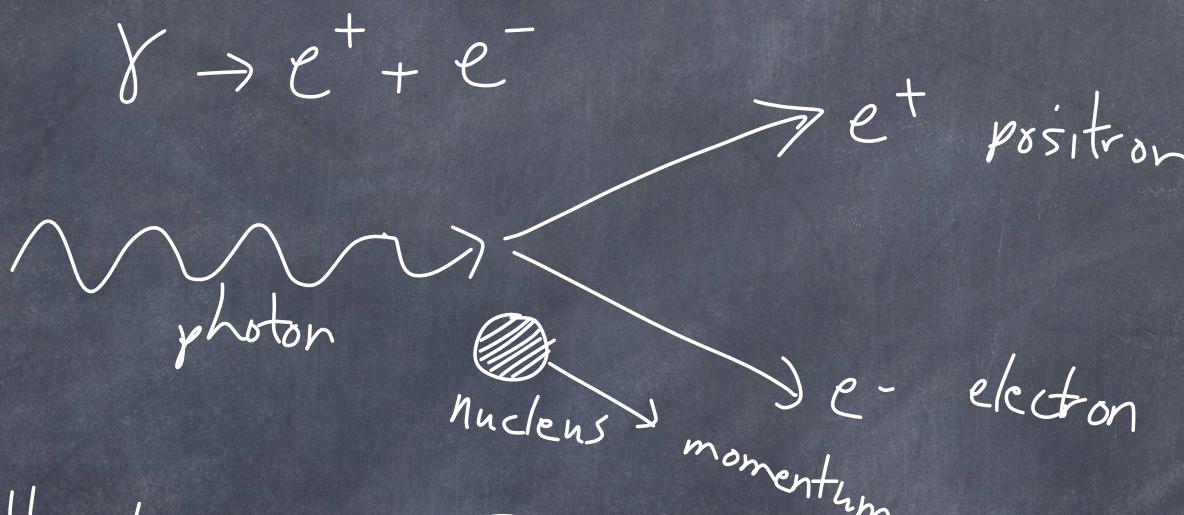


x-rays



gamma rays

IF a photon has enough energy, it can convert its energy entirely to charged particles. The lightest charged particle is the electron.



This can't happen in free space, only near a massive object, such as a nucleus, such that the nucleus can supply momentum so that the momentum is conserved.

How much photon energy is enough?

$$E = h\nu \geq 2m_e c^2 \quad m_e = 0.511 \text{ MeV}/c^2$$
$$E > 1.022 \text{ MeV}$$

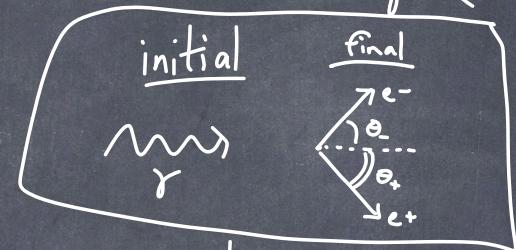
start S3: Supplementary proof that $\gamma \rightarrow e^+ + e^-$ can't happen in free space

we assume no nucleus:

initial: photon has energy: $h\nu$

final: no photon anymore,
electron has energy E_- and momentum \vec{p}_-
positron has energy E_+ and momentum \vec{p}_+

$m = m_e$: mass of electron



From conservation laws:

initial final

$$\text{Energy: } h\nu = E_+ + E_- \quad \textcircled{1}$$

$$\text{Momentum } x: \frac{h\nu}{c} = p_- \cos\theta_- + p_+ \cos\theta_+ \quad \textcircled{2}$$

$$\text{momentum } y: 0 = p_- \sin\theta_- + p_+ \sin\theta_+ \quad \textcircled{3}$$

Rewrite $\textcircled{2}$ we get: $h\nu = cp_- \cos\theta_- + cp \cos\theta_-$ $\textcircled{4}$

Insert formula for relativistic energy ($\epsilon^2 = (\epsilon p)^2 + (mc^2)^2$) into $\textcircled{1}$:

$$h\nu = \sqrt{(\epsilon p_+)^2 + (mc^2)^2} + \sqrt{(\epsilon p_-)^2 + (mc^2)^2} \quad \textcircled{5}$$

The maximum value of $h\nu$ in ④ is when
 $\cos\theta_- = \cos\theta_+ = 1$.

Then ④ becomes
$$h\nu = c p_- + c p_+ \quad ⑤$$

But if we look at eq. ⑤, we see that
 $(h\nu)^2$ must be greater than $(cp_-)^2 + (cp_+)^2$
because of the electron + positron masses.

Therefore, since we have 2 equations, ⑤ and ⑥,
which can't be both true at the same time,
this reaction is not valid, because energy &
momentum can't be conserved simultaneously.

end

⑤ ⑥ finished

what happens to positrons ?

\oplus \ominus They orbit each other
(positron exists for about
 10^{-10} s)

Then they annihilate



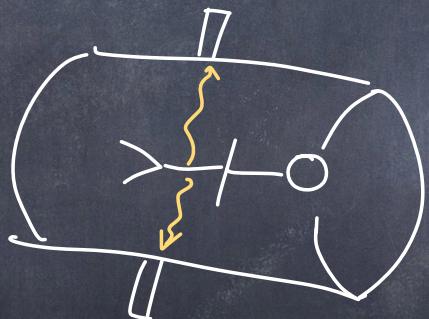
(Two photons instead of one is
because of momentum & energy
conservation)

This is the principle of Positron emission tomography (PET). we start with a positron.

- 1) A positron-emitting radioactive element containing ^{15}O , ^{11}C , ^{13}N , ^{18}F , ^{68}Ga is attached to a pharmaceutical + (ingested or injected) Radioactive elements are usually prepared at an accelerator.

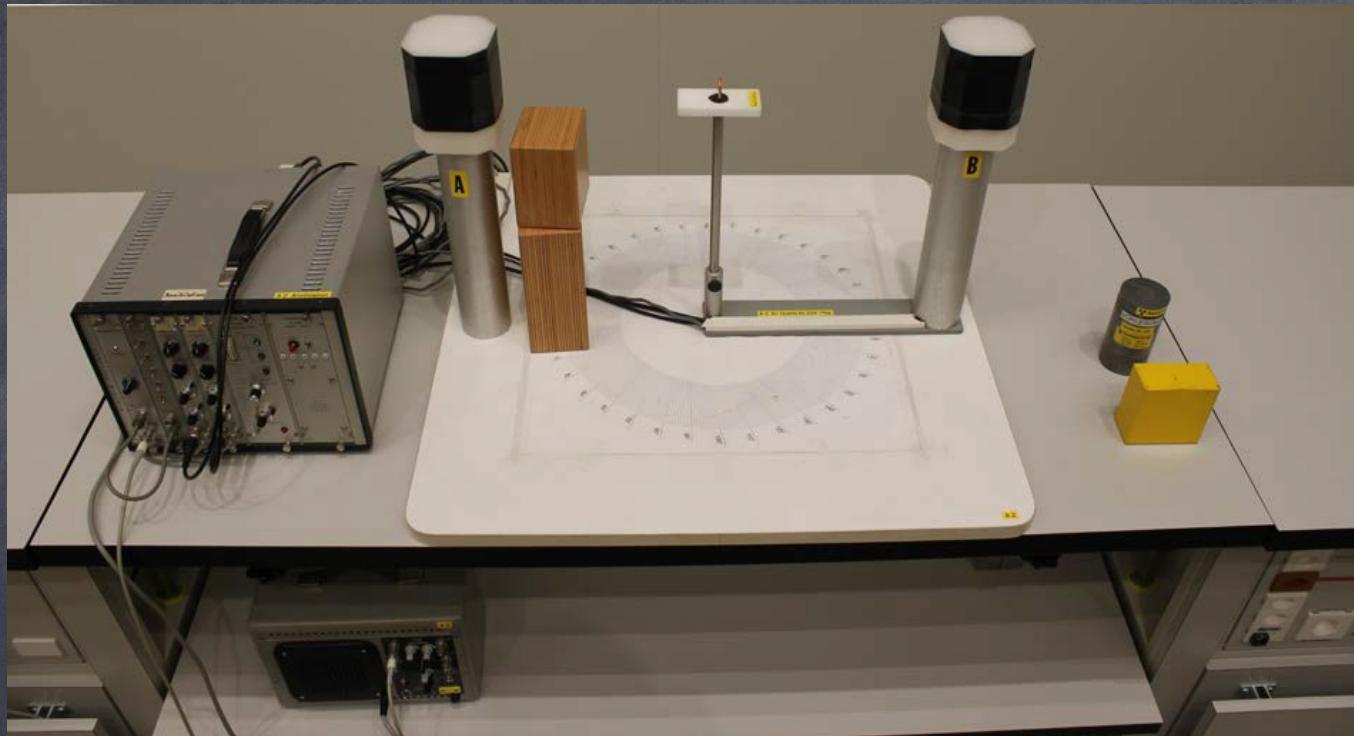
This is the principle of Positron emission tomography (PET).

- i) A positron-emitting radioactive element containing ^{15}O , ^{11}C , ^{13}N , ^{18}F , ^{68}Ga is attached to a pharmaceutical + (ingested or injected) Radioactive elements are usually prepared at an accelerator.

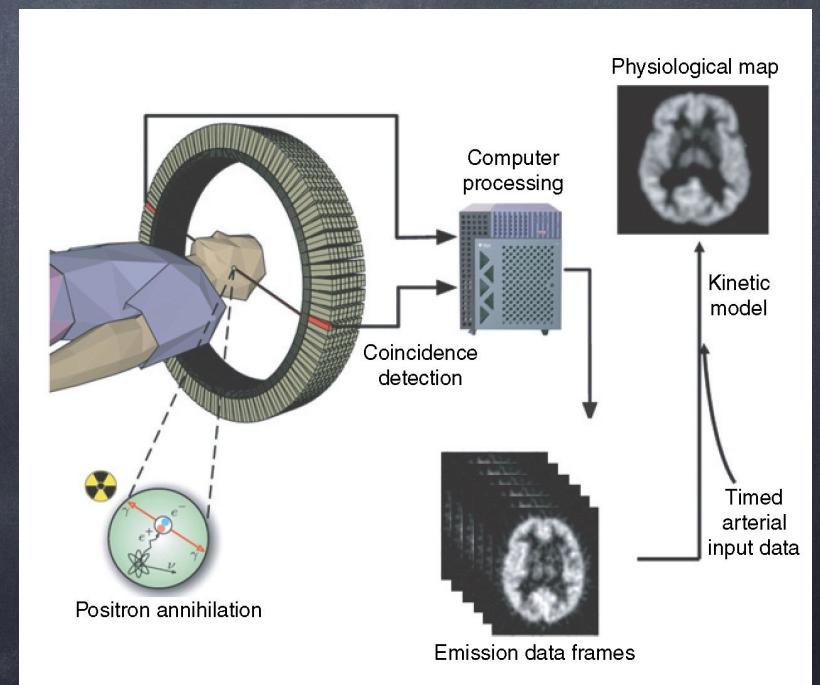
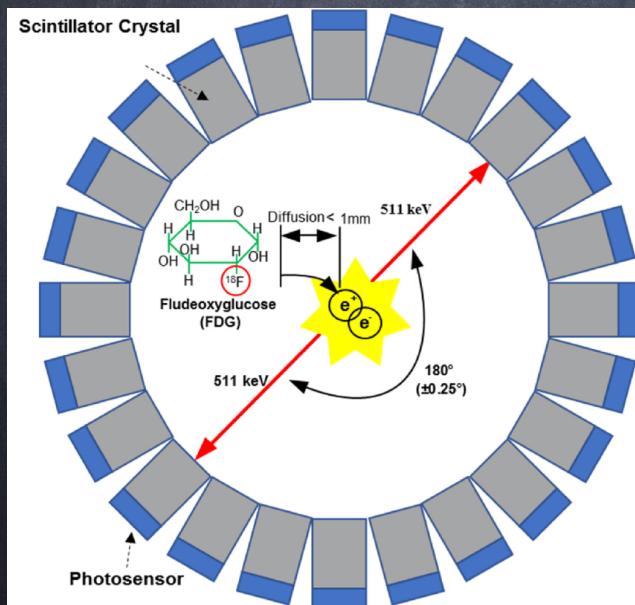
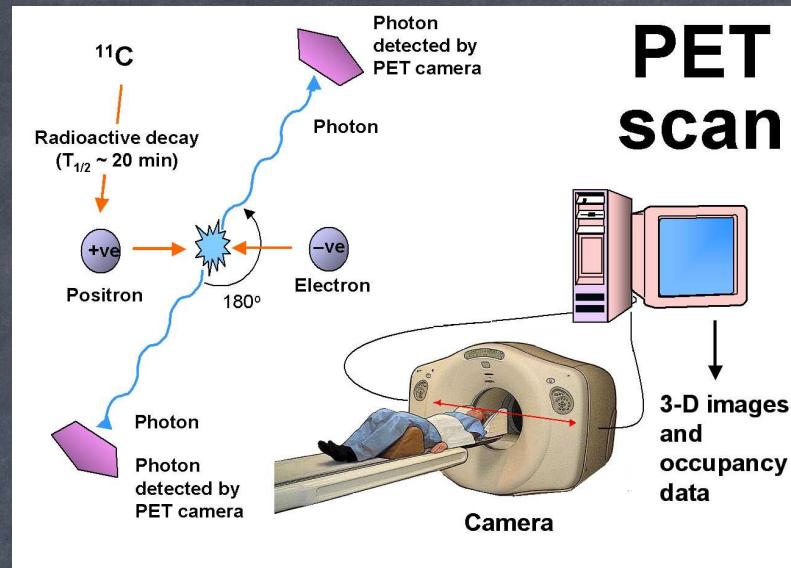


Two photons leave the body back to back. Detectors 180° apart that look for coincident arrival of 511 keV gamma rays.

By collecting data at different angles, we can reconstruct 3D images.



PET scan



Spatial resolution limited to $\sim 5\text{mm}$, by:

- 1) positronium has some non-zero momentum,
So the angle is not exactly 180°
- 2) positron can travel $\sim 1\text{ mm}$ before
it annihilates.

But PET scans can be done in real time.

By correlating images of blood flow or glucose or oxygen metabolism, & monitoring a patient stimulated in some way, biochemical events can be correlated with brain activity.
(Can reveal abnormal brain function)

PET is best used for monitoring time dependence on metabolism of radiopharmaceuticals, but not the best technique for spatial resolution.

Supplement on Feynman
diagrams follows

Light is an electromagnetic wave.

Light is quantized. The unit of light is a photon.

Quantum electrodynamics (QED)

All electromagnetic phenomena are ultimately reducible to the following elementary process.

(with
an
electron)



Time flowing horizontally to the right
This diagram reads "an electron enters, emits, or absorbs a photon, and exits."

This diagram can be flipped or rotated,
and the process still happens.



A particle moving backwards in time is interpreted
as an antiparticle moving forwards in time.

$$\text{electron} = e^-$$

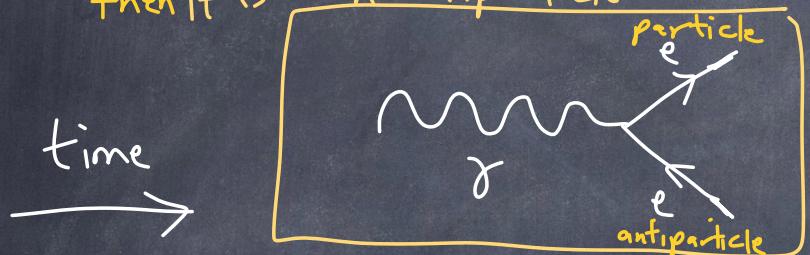
positron = e^+ antiparticle of the electron

A photon does not need an arrow since it is
its own antiparticle.

So this diagram reads "a positron enters,
emits or absorbs a photon,
and exits."

The positron was predicted in 1928
by Dirac because his formals had 2 solutions: +,-

If arrow is moving opposite to time, Discovered in 1932 by Anderson
then it is an antiparticle.



Can happen but must
obey energy & momentum
conservation.

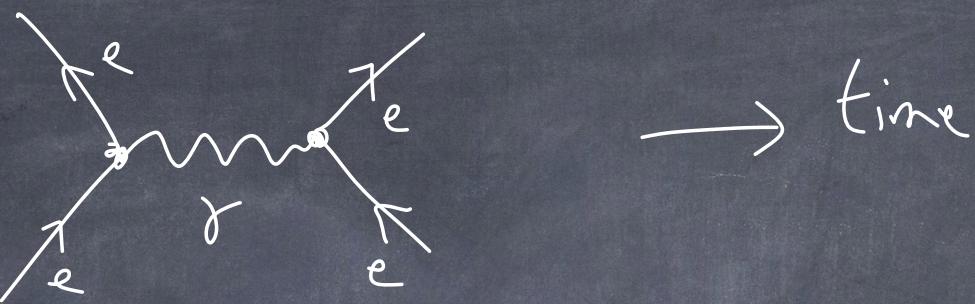
"A photon enters, decays into an electron
and a positron, and they exit."

These diagrams are called Feynmann diagrams.



Some people label these diagrams
with $e^- + e^+$, but I find
this dangerous.

Since a positron moving backwards
in time would be an electron



Here, an electron and positron annihilate into a photon, and then the photon decays into a new electron and positron.

Note: the electric charge is conserved.

We can write this diagram as:

$$e^- + e^+ \rightarrow \gamma \rightarrow e^- + e^+$$

electric charge $-1 + 1 = 0 = 0 = -1 + +1$

Energy & momentum are conserved in this process.

This diagram can be rotated.



time
→

Here, two electrons enter,
exchange a photon
and continue as
electrons.

Here, the electrons repel can be seen to repel
each other.

In quantum physics, forces
are mediated by particles.

The photon mediates the
electromagnetic force.

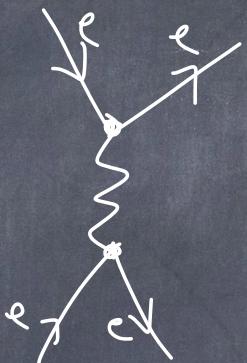
But does this mean classical physics is wrong?

classical physics
view:

$$-\bar{F}_{21} = \bar{F}_{12} = \frac{k q_1 q_2}{r_{12}^2}$$

Electrons are repelled
by a force, which
we can calculate.

(PHY 117) quantum physics view: time



Here, two electrons
exchange a photon
and continue as
electrons.

Here, the electrons repel
each other.

In quantum physics, forces
are mediated by particles.

The photon mediates the
electromagnetic force.

In practice, classical physics is easier to
calculate for most everyday situations

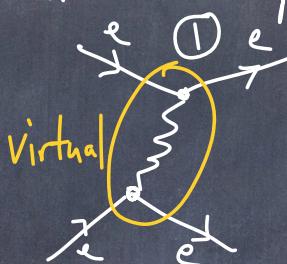
What happens if a particle moves perpendicular to time?

A:

Here a photon moves vertically.

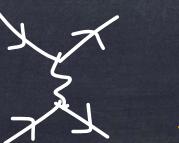


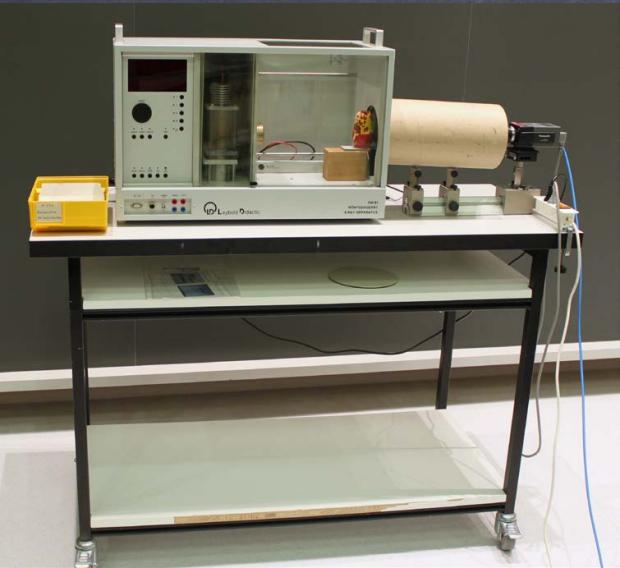
Really, what is happening is both of these:



In ①, the photon is emitted from the below electron.
In ②, the photon is emitted from the above electron

The photon is not observable, we call it virtual.
we can't tell if ① or ② happens, so we
use quantum mechanics to consider both.
But we draw it like this

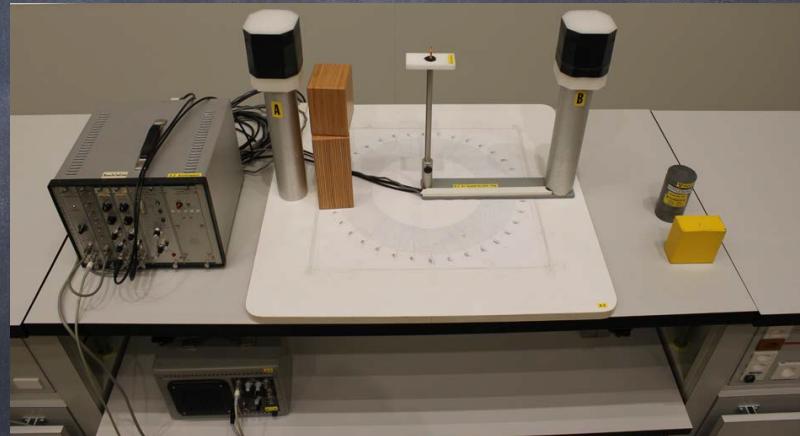




A24



ES90



A2